Nuclear Chemistry – Alpha, Beta, and Gamma Decay Practice

Identify the following examples as alpha, beta, or gamma decay.

1.
$$^{233}_{92}U \rightarrow ^{4}_{2}He + ^{229}_{90}Th _alpha_$$

4.
$$^{202}_{81}\text{TI} \rightarrow ^{202}_{81}\text{TI} + ^{0}_{0}\text{Y}$$
 gamma

2.
$$^{99}_{43}$$
Tc $\rightarrow ^{0}_{-1}$ e + $^{99}_{44}$ Ru beta

s.
$${}^{125}_{53}I \rightarrow {}^{0}_{0} \Upsilon + {}^{125}_{53}I$$
 gamma

3.
$$^{175}_{78}$$
 Pt $\rightarrow ^{171}_{76}$ Os + $^{4}_{2}$ He _alpha_

6.
$${}^{82}_{35}Br \rightarrow {}^{82}_{36}Kr + {}^{0}_{-1}e$$
 __beta___

Alpha Decay: Write the alpha decay equations for the following nuclides. Example: ${}^{208}_{84}$ Po $\rightarrow {}^{4}_{2}$ He ${}^{204}_{82}$ Pb

Example:
$${}^{208}_{84}$$
 Po $\rightarrow {}^{4}_{2}$ He + ${}^{204}_{82}$ Pb

7.
$$^{256}_{103}$$
Lr $\rightarrow ^{4}_{2}$ He + $^{252}_{101}$ Md

s.
$$^{231}_{91}$$
 Pa $\rightarrow ^{4}_{2}$ He + $^{227}_{89}$ Ac

Beta Decay: Write the beta decay equations for the following nuclides.

Example:
$${}^{14}_{6}C \rightarrow {}^{14}_{7}N + {}^{0}_{-1}e$$

11.
$$^{24}_{11}$$
 Na $\rightarrow ^{24}_{8}$ O + $^{0}_{-1}$ e

13.
$$^{247}_{95}$$
 Am $\rightarrow ^{247}_{96}$ Cm + $^{0}_{-1}$ e

12.
$$^{201}_{79}$$
 Au $\rightarrow ^{201}_{80}$ Hg + $^{0}_{-1}$ e

14.
$$^{90}_{38}$$
Sr $\rightarrow ^{90}_{39}$ Y + $^{0}_{19}$ e

Balancing Decay Reactions: Fill in the blank in each of the following decay reactions with the correct decay particle or decayed nucleus that will balance the reaction, also state whether it is alpha or beta decay

15.
$$^{184}_{74}$$
 W $\rightarrow ^{4}_{2}$ He + $^{180}_{72}$ Hf

20.
$$^{60}_{27}W \rightarrow ^{0}_{-1}e + ^{60}_{28}Ni$$

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- 1. Understanding the eBook Nuclear Chemistry Practice Exams
 - The Rise of Digital Reading Nuclear Chemistry Practice Exams
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Nuclear Chemistry Practice Exams
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nuclear Chemistry Practice Exams
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Nuclear Chemistry Practice Exams

- Personalized Recommendations
- Nuclear Chemistry Practice Exams User Reviews and Ratings
- Nuclear Chemistry Practice Exams and Bestseller Lists
- 5. Accessing Nuclear Chemistry Practice Exams Free and Paid eBooks
 - Nuclear Chemistry Practice Exams Public Domain eBooks
 - Nuclear Chemistry Practice Exams eBook Subscription Services
 - Nuclear Chemistry Practice Exams Budget-Friendly Options
- 6. Navigating Nuclear Chemistry Practice Exams eBook Formats
 - o ePub, PDF, MOBI, and More
 - Nuclear Chemistry Practice Exams Compatibility with Devices
 - Nuclear Chemistry Practice Exams Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nuclear Chemistry Practice Exams
 - Highlighting and Note-Taking Nuclear Chemistry Practice Exams
 - Interactive Elements Nuclear Chemistry Practice Exams
- 8. Staying Engaged with Nuclear Chemistry Practice Exams
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nuclear Chemistry Practice Exams
- 9. Balancing eBooks and Physical Books Nuclear Chemistry Practice Exams
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nuclear Chemistry Practice Exams
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nuclear Chemistry Practice Exams
 - Setting Reading Goals Nuclear Chemistry Practice Exams
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nuclear Chemistry Practice Exams

- Fact-Checking eBook Content of Nuclear Chemistry Practice Exams
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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