

A chemical bond - forms when 2 or more atoms rearrange _____
_____ to increase _____.

ionic bond - forms when valence _____ are _____
from one atom to another.

cation - atom _____ electrons to become _____ charged.

anion - atom _____ electrons to become _____ charged.

In ionic compounds the ions are arranged in a _____
_____ forces hold the ions together.

Properties of ionic compounds:

- High _____ and _____ points
- _____ - not easily _____
- _____ electricity when _____ or
_____ because the ions are free to _____.

covalent bond - _____ are _____, forming
_____.

Covalent compounds have _____ forces holding the
_____ together.

Properties of covalent compounds:

- lower _____ and _____ points
- Many covalent compounds are _____ liquids or gases.
- _____ - easier to _____
- are not _____ of electricity

electronegativity - property that tells how strong an atom's _____
is for _____.

Since oxygen has a _____ electronegativity than hydrogen, oxygen
holds onto shared electrons _____, giving the oxygen a _____
negative charge and the hydrogen a partial _____ charge.

Note Taking Guide Episode 501

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