

CRUDE OIL

Task 1 - Crude oil key word definitions:

Crude oil is a finite resource found in rocks. It is the remains of plant & animal life that was buried in mud many millions of years ago.

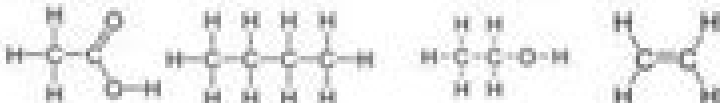
a) Define finite resource

Crude oil is a mixture of a very large number of compounds. Most of the compounds in crude oil are hydrocarbons.

Define Hydrocarbon

Task 2 - Hydrocarbons:

a) Highlight which of the following are not hydrocarbons. Explain your answer



Task 3 - Alkanes:

Alkanes are the simplest type of hydrocarbon. They contain only single bonds & have the general formula: $\text{C}_n\text{H}_{2n+2}$. Complete the following table of alkanes:

Name	Molecular formula	Displayed formula
methane	CH_4	$\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{H} \\ \\ \text{H} \end{array}$
_____	C_2H_6	
propane	C_3H_8	
butane		

Task 4 - Fractional distillation:

a) What is the process of fractional distillation used for?

b) Add labels to the diagram opposite to make clear what happens as you go up the column to the following:

- Size of molecule
- Boiling point of hydrocarbon
- Viscosity of hydrocarbon
- Flammability

c) What property of the molecules does fractional distillation rely upon to separate them from crude oil?

d) Bitumen is collected at the bottom of the column. It is still a liquid & did not turn into a vapour when heated. What does that tell you about the boiling point of bitumen compared to the rest of the hydrocarbons in crude oil?

e) Choose the correct word from the word bank to complete this description of the process of fractional distillation:

Fractional distillation separates the crude oil into different _____ according to the _____ of the different hydrocarbons.

So each fraction contains a mixture of different-sized hydrocarbons with similar boiling points.

The longer molecules will have greater _____ of attraction between them & so will need more _____ to overcome these forces. This means these _____ chain hydrocarbons will boil at _____ temperatures.

Intermolecular forces	higher	fractions
boiling points	energy	longer

Task 5 - Combustion:

Hydrocarbons are used as fuels as they release a lot of energy when they are burnt.

a) Write the word equation to show the combustion of a hydrocarbon fuel:

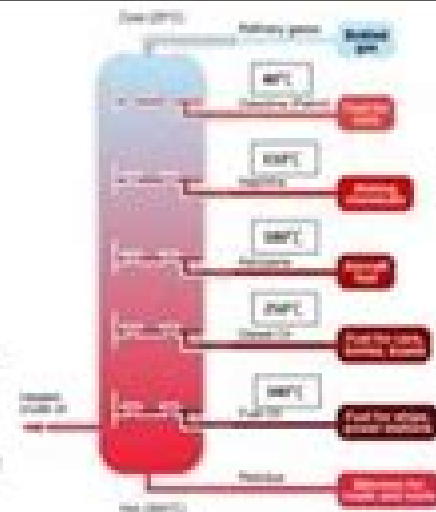
HYDROCARBON + _____ → _____ + _____

b) Following is the equation for the combustion of methane gas, CAN YOU BALANCE IT?



c) What do we call the reaction that happens when we try to burn a fuel without enough oxygen?

d) What is the toxic gas that is produced by this chemical reaction? Explain why it is harmful.



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Table of Contents Ocr Gcse Chemistry C7 Mark Scheme

1. Understanding the eBook Ocr Gcse Chemistry C7 Mark Scheme
 - The Rise of Digital Reading Ocr Gcse Chemistry C7 Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Gcse Chemistry C7 Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Gcse Chemistry C7 Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Gcse Chemistry C7 Mark Scheme
 - Personalized Recommendations
 - Ocr Gcse Chemistry C7 Mark Scheme User Reviews and Ratings
 - Ocr Gcse Chemistry C7 Mark Scheme and Bestseller Lists
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 - Ocr Gcse Chemistry C7 Mark Scheme Public Domain eBooks
 - Ocr Gcse Chemistry C7 Mark Scheme eBook Subscription Services
 - Ocr Gcse Chemistry C7 Mark Scheme Budget-Friendly Options

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 - Ocr Gcse Chemistry C7 Mark Scheme Compatibility with Devices
 - Ocr Gcse Chemistry C7 Mark Scheme Enhanced eBook Features
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 - Highlighting and Note-Taking Ocr Gcse Chemistry C7 Mark Scheme
 - Interactive Elements Ocr Gcse Chemistry C7 Mark Scheme
8. Staying Engaged with Ocr Gcse Chemistry C7 Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Gcse Chemistry C7 Mark Scheme
9. Balancing eBooks and Physical Books Ocr Gcse Chemistry C7 Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Gcse Chemistry C7 Mark Scheme
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr Gcse Chemistry C7 Mark Scheme
 - Setting Reading Goals Ocr Gcse Chemistry C7 Mark Scheme
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr Gcse Chemistry C7 Mark Scheme
 - Fact-Checking eBook Content of Ocr Gcse Chemistry C7 Mark Scheme
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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