

Chemistry C4-Chemical Economics

- 1. The pH scale and chemical modification:



pH	Chemical	
Increasingly acidic	0-1	strongest acids
	1-2	strong acids
	2-3	strong acids
	3-4	strong acids
	4-5	weak acids
	5-6	weak acids
	7	neutral
Neutral	8	weak bases
	9	weak bases
	10	strong bases
	11	strong bases
	12	strong bases
	13	strong bases
	14	strongest bases
Increasingly alkaline	15	strongest bases

- 2. The modification is just a step that changes colour:
 - a) This step in modification changes colour depending on the pH
 - b) Chemical modification is a very useful modification of dye which gives the colour change
 - c) It is useful for monitoring the pH of a solution
- 3. The pH is a measure of the concentration of H⁺ ions in a solution:
 - a) The pH is a measure of the concentration of H⁺ ions in a solution
 - b) The pH is a measure of the concentration of H⁺ ions in a solution
 - c) The pH is a measure of the concentration of H⁺ ions in a solution
- 4. The pH is a measure of the concentration of H⁺ ions in a solution:
 - a) The pH is a measure of the concentration of H⁺ ions in a solution
 - b) The pH is a measure of the concentration of H⁺ ions in a solution
 - c) The pH is a measure of the concentration of H⁺ ions in a solution

Ocr Gateway Chemistry 2013 C4 5 6

Vicki Wright



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