## 4.2.4 Spectroscopy

Specification reference	Checklist questions
72.4 a	Can you describe how infrared (IR) radiation causes covalent bonds to vibrate more and absorb energy?
4.2.4 6	Can you explain absorption of infrared radiation by atmospheric gases containing C=O, O=H and C=H bonds (e.g. H <sub>2</sub> O, CO <sub>2</sub> and CH <sub>4</sub> ), the suspected link to global warming and resulting changes to energy usage?
72.4 c i	Can you use an infrared spectrum of an organic compound to identify an alcohol from an absorption peak of the O-H bond?
7.2.4 e ii	Can you use an infrared spectrum of an organic compound to identify an aldehyde or ketone from an absorption peak of the C=O bond?
	Can you use an infrared spectrum of an organic compound to identify a carboxylic acid from an absorption peak of the C=O bond and a broad absorption peak of the O-H bond?
42.4 d	Can you interpret and predict an infrared spectrum of familiar or unfamiliar substances using supplied data?
4.2.4 e	Can you use infrared spectroscopy to monitor gases causing air pollution (e.g. CO and NO from car emissions) and in modern breathalysers to measure ethanol in the breath?
A.2.4 f	Can you use a mass spectrum of an organic compound to identify the molecular ion peak and hence to determine molecular mass?
4.2.4 9	Can you analyse fragmentation peaks in a mass spectrum to identify parts of structures?
4.2.4 h i	Can you deduce the structures of organic compounds from different analytical data including elemental analysis?
3.2.4 h ii	Can you deduce the structures of organic compounds from different analytical data including mass spectra?

# Ocr Gateway End Of Unit Test Oxford

**Stephen Pople** 

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