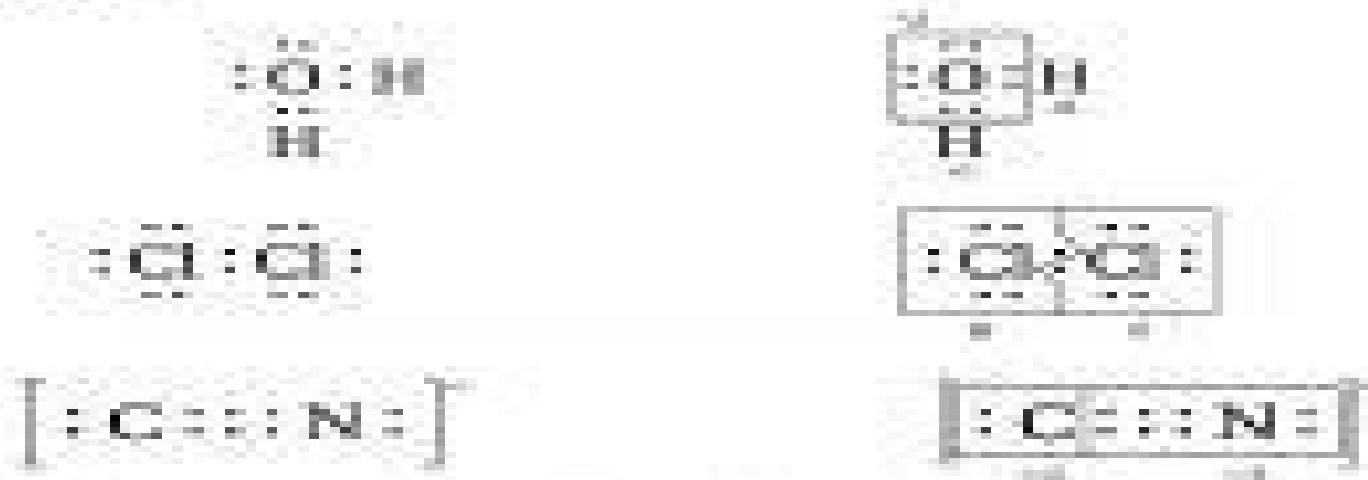


Resonance-Structural Drawing

Resonance Structures:



1. correct electron-distribution, the number of valence electrons, and the overall charge of the molecule, correct placement of lone pairs

These Lewis structures show the same:

1) Same atom positions

2) Same total # electrons for neutral molecules (e.g., 10)

3) A structure with a double bond on the right side of the molecule will be an equivalent structure if the double bond is located on the left side.



2nd set

These Lewis structures show differences:

1) Different atom positions are

2) Different # electrons for ionic species



3rd set

These are 3 completely different molecules (e.g., 8, 10, 12)

1) Same atom positions, 2) 8

3) Different # of lone pairs (electron pairs) on the atoms

4) Overlapped atoms (e.g., 10) will be the same molecule (e.g., 10)

5) Overlapped atoms (e.g., 12) will be the same molecule (e.g., 12)



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CL Gary

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'CL Gary' text.

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