














# VSEPR Geometries

# of electron groups	Types of electron groups	VSEPR	Name of molecular shape	Ex
2	2 Bonds	$AX_2$	Linear 	$BeF_2$
3	3 Bonds	$AX_3$	Trigonal planar 	$BF_3$
3	2 Bonds, 1 Lone Pair	$AX_2E$	Angular 	$SnCl_2$
4	4 bonds	$AX_4$	Tetrahedral 	$CF_4$
4	3 bonds, 1 lone pair	$AX_3E$	Trigonal pyramidal 	$PCl_3$
4	2 bonds, 2 lone pair	$AX_2E_2$	Angular 	$H_2S$
5	5 bonds	$AX_5$	Trigonal bipyramidal 	$SbCl_5$
5	4 bonds, 1 lone pair	$AX_4E$	Seesaw 	$TeCl_4$
5	3 bonds, 2 lone pair	$AX_3E_2$	T-shaped 	$BrF_3$
5	2 bonds, 3 lone pair	$AX_2E_3$	Linear 	$XeF_2$
6	6 bonds	$AX_6$	Octahedral 	$SF_6$
6	5 bonds, 1 lone pair	$AX_5E$	Square pyramidal 	$BrF_5$
6	4 bonds, 2 lone pair	$AX_4E_2$	Square planar 	$XeF_4$

# Molecular Geometry Study Guide

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