

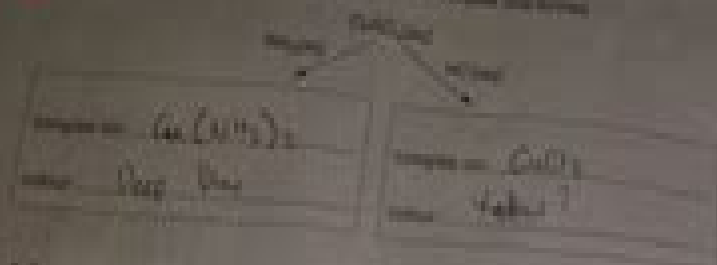
Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes and ligands.

(a) The reaction shows the synthesis of a copper complex.

In the boxes, write the formula and name of the ligands and metal.



(b) Cu^{2+} has two d-orbitals in d_{sp^2} hybridization state and two water molecules. The d_{sp^2} hybridization is a tetrahedral ligand.

The resulting formula of the complex is shown in Fig. 1.1 below.



Fig. 1.1

(c) What is meant by the term coordination bond?

A bond formed between a central metal ion and a ligand.

(d) The reaction is shown below.

The reaction is about the synthesis of a copper complex. The charge has been given.



Complex A

Complete the d_{sp^2} diagram of the other two geometries of A. You do not need to include any charges.

Indicate with ticks whether the geometries are oct, tetra, square or a combination of these types.

Diagram														
Type	<table border="1"> <tr> <td>oct</td> <td></td> </tr> <tr> <td>tetra</td> <td><input checked="" type="checkbox"/></td> </tr> <tr> <td>square</td> <td></td> </tr> </table>	oct		tetra	<input checked="" type="checkbox"/>	square		<table border="1"> <tr> <td>oct</td> <td><input checked="" type="checkbox"/></td> </tr> <tr> <td>tetra</td> <td></td> </tr> <tr> <td>square</td> <td><input checked="" type="checkbox"/></td> </tr> </table>	oct	<input checked="" type="checkbox"/>	tetra		square	<input checked="" type="checkbox"/>
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(e) What is the empirical formula, including the charge, of the complex in A?



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