

Name: _____ Date: _____ Lab Section _____

Experiment 2: Molecular Models REPORT SHEET

Pre-lab Study Questions:

1. Objectives of the experiment (in your own words):

2. What kinds of electrons form the covalent bonds in molecules?

3. Is a polar molecule symmetrical or asymmetrical?

FORMULA OF MOLECULE	LEWIS STRUCTURE	3-D SHAPE w/lone pairs	MOLECULAR GEOMETRY	MOLECULAR POLARITY
example: CH ₄	$\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{H} \\ \\ \text{H} \end{array}$		tetrahedral	nonpolar
H ₂ O	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad / \\ \text{O} \\ / \quad \backslash \\ \text{H} \quad \text{H} \end{array}$	$\text{H} \quad \text{H} \quad \text{O} \quad \cdot \cdot$	tetrahedral	polar
H ₂ O ⁺	$\text{H} \quad \text{H} \\ \diagdown \quad / \\ \text{O} \\ / \quad \backslash \\ \text{H} \quad \text{H}$	$\text{H} \quad \text{H} \quad \text{O} \quad \cdot \cdot$	tetrahedral	nonpolar
C ₂ H ₆				
C ₂ H ₄	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad / \\ \text{C} = \text{C} \\ / \quad \backslash \\ \text{H} \quad \text{H} \end{array}$		trigonal planar	non polar
C ₂ H ₂				
CO ₂	$\text{O} = \text{C} = \text{O}$		Linear	Non polar
H ₂ O ₂				
CHCl ₃				

Molecular Model Laboratory Manual Answers

David Wright Wilson



Molecular Model Laboratory Manual Answers:

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Study Guide and Solutions Manual Neil E. Schore, 2002-08-02
Practical Manual Fundamentals of Plant Biochemistry and Biotechnology Dr. Dinesh Kumar, Dr. Chandra Gurnani, Dr. Priyanka Sharma, 2025-04-15 This manual is designed to provide a detailed and practical guide for students researchers and

practitioners involved in the study of biochemistry molecular biology and plant tissue culture The topics covered herein are fundamental to the understanding and application of laboratory techniques and processes used in a variety of biological and biochemical studies The manual starts with the preparation of solutions pH adjustment and the use of buffers essential skills in any biological laboratory It then progresses through qualitative tests for carbohydrates and amino acids quantitative estimations of glucose and proteins and titration methods for amino acids and lipids providing a comprehensive overview of common biochemical assays These methods are critical for gaining insights into the molecular composition and behavior of biological samples Special focus is placed on enzyme kinetics and how factors such as pH temperature and substrate concentration influence enzyme activity concepts that are vital in both research and applied biochemistry Additionally techniques like paper chromatography and thin layer chromatography TLC for separating amino acids and monosaccharides are explored giving readers practical skills for analyzing and identifying complex biomolecules The manual also addresses the increasingly important field of plant biotechnology introducing sterilization techniques tissue culture media composition and the preparation of stock solutions for Murashige and Skoog MS nutrient medium It covers callus induction micro propagation and the processes of hardening and acclimatization which are essential for producing genetically uniform plantlets in vitro Moreover the manual provides demonstrations on advanced molecular techniques such as DNA isolation gel electrophoresis and DNA fingerprinting tools that are indispensable for genetic studies and molecular diagnostics By compiling these diverse yet interrelated techniques this manual aims to equip readers with a solid foundation in both traditional and cutting edge laboratory practices Whether used in educational settings or research laboratories this manual serves as an invaluable resource for mastering the essential techniques of modern biological science

Comprehensive Practical Manual of Pharmaceutical Chemistry Shaik Munwar, Shaik Ammaji, 2019-02-28 The edition of Comprehensive Practical Manual of Pharmaceutical Chemistry is authored in simple and comprehensive style according to PCI Pharmacy Council of India syllabus to meet the specific needs of the pharmacy students It provides comprehensive yet concise chemistry for D Pharmacy B Pharmacy M Pharmacy and Pharm D students The main objective of this manual is to attract students to learn the basic theories of pharmaceutical chemistry thus the manual is aimed to enrich the inadequacy in teaching and learning of pharmaceutical chemistry by providing enormous information The style of presentation of this manual is such that it not only gives deeper understanding of the subject but also will help the beginners to overcome the fright of the subject The manual gives concise and pointwise information required during practicals in single book and eliminates the need of too many reference books during practicals The manual authored in simple lucid and easy language

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that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives Laboratory Manual of Inorganic Chemistry John Bernard Ekeley,1923

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