

1. What are the four most commonly occurring elements in organisms?

Carbon, Hydrogen, Nitrogen, Oxygen

2. Define organic molecule.

A molecule that contains carbon (with the exception of carbonates -  $\text{CO}_2$  and -  $\text{CO}_3$ )

3. Why is carbon the ideal building block for large molecules?

Carbon can make covalent bond with up to four other atoms

4. Define the following terms:

a. Metabolism:

The overall network of chemical reactions that occur in cells

b. Catabolism:

The set of reactions that break larger molecules down to smaller ones

c. Anabolism:

The set of reactions that assemble large molecules from smaller parts

5. Outline the theory of Vitalism and how Wohler's synthesis of Urea disproved the theory:

Vitalism was a theory that organic compounds could only be made by living things due to their "vital force." When Wohler was able to produce Urea (an organic compound) artificially in his lab it showed that organic compounds could still be formed without the "vital force" of life.

6. Complete the table of common organic molecule groups / parts:

Name of Group	Hydroxyl	Amine	Carboxyl	Methyl
Full Structure	$\text{S}-\text{O}-\text{H}$	$\text{S}-\text{N}-\text{H}$ $\quad \quad \quad \text{H}$	$\text{S}-\text{C}-\text{OH}$ $\quad \quad \quad \text{O}$	$\text{S}-\text{C}-\text{H}$ $\quad \quad \quad \text{H}$ $\quad \quad \quad \text{H}$
Simplified Notation	$-\text{OH}$	$-\text{NH}_2$	$-\text{COOH}$	$-\text{CH}_3$

7. Complete the table of common organic molecules:

Name	Ribose	Glucose	Saturated Fatty Acid	Amino Acid
Drawing				

8. Describe the structure a single water molecule and how this leads to its dipolar (polar) nature:

Two Hydrogen atoms are bonded to one Oxygen atom by covalent bonds. However, electrons in these bonds are not shared evenly so the Oxygen takes on a partial negative charge and Hydrogen takes

9. In the space below, draw three water molecules attracted to one another by hydrogen bonding. Include labels to show the polarity of the molecules and the hydrogen bonds:



a  
partial  
positive  
charge

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## **Table of Contents Molecules By Acme Answer Key**

1. Understanding the eBook Molecules By Acme Answer Key
  - The Rise of Digital Reading Molecules By Acme Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Molecules By Acme Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Molecules By Acme Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Molecules By Acme Answer Key
  - Personalized Recommendations
  - Molecules By Acme Answer Key User Reviews and Ratings

- Molecules By Acme Answer Key and Bestseller Lists
- 5. Accessing Molecules By Acme Answer Key Free and Paid eBooks
  - Molecules By Acme Answer Key Public Domain eBooks
  - Molecules By Acme Answer Key eBook Subscription Services
  - Molecules By Acme Answer Key Budget-Friendly Options
- 6. Navigating Molecules By Acme Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Molecules By Acme Answer Key Compatibility with Devices
  - Molecules By Acme Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Molecules By Acme Answer Key
  - Highlighting and Note-Taking Molecules By Acme Answer Key
  - Interactive Elements Molecules By Acme Answer Key
- 8. Staying Engaged with Molecules By Acme Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Molecules By Acme Answer Key
- 9. Balancing eBooks and Physical Books Molecules By Acme Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Molecules By Acme Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Molecules By Acme Answer Key
  - Setting Reading Goals Molecules By Acme Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Molecules By Acme Answer Key
  - Fact-Checking eBook Content of Molecules By Acme Answer Key
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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