

# SHOP MANUAL

## MASSEY-FERGUSON

### MODELS

362 365 375 383 390 390T 398

The tractor serial number is located in two locations, on the serial number plate (Fig. 1) located on the right side of the tractor and stamped on the rear axle casting (Fig. 2). The tractor serial number is coded to identify the machine type, year of manufacture, week of manufacture (during specific year) and specific unit during week.

The illustration shown in Fig. 3 identifies a two-wheel-drive 390 tractor which was the 121st tractor manufactured during the 32nd week of 1988.

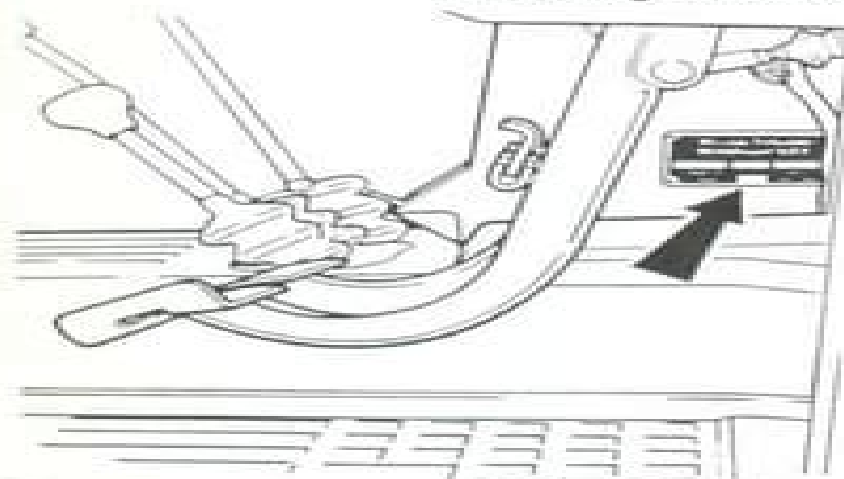


Fig. 1—The tractor serial number is located on plate attached to tractor right side as shown as well as stamped on axle housing as shown in Fig. 2.

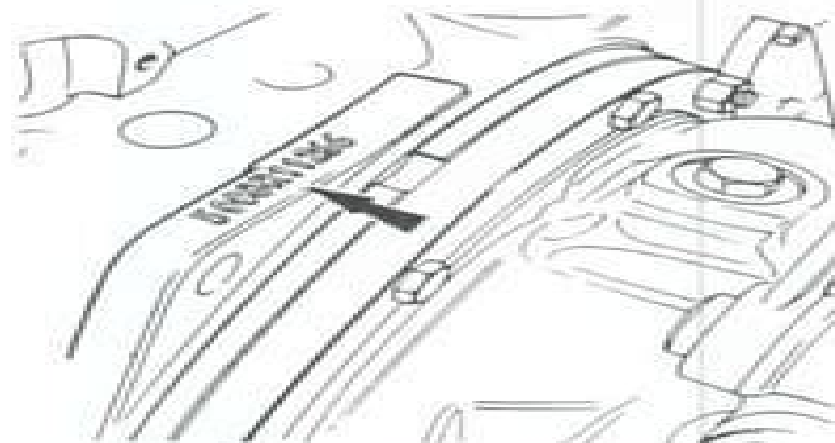


Fig. 2—Tractor serial number is stamped into rear axle housing as shown as well as on plate shown in Fig. 1. Numbers match on original assembly.

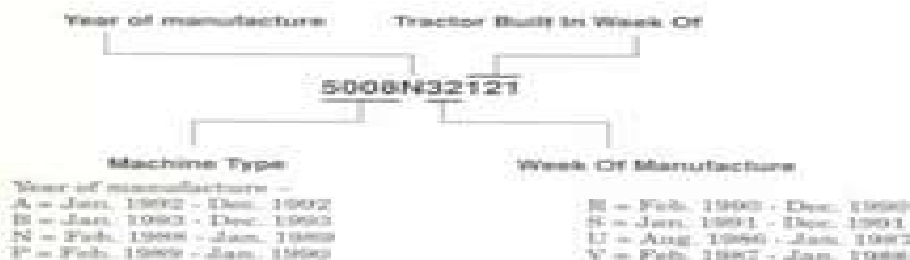


Fig. 3—The tractor serial number identifies machine type, year of manufacture, week of manufacture during specific year, and specific unit manufactured during that week. The example shown identifies a two-wheel-drive 390 tractor which was the 121st tractor manufactured during the 32nd week of 1988.

Machine type —  
 5006 M-F 375 2WD  
 5007 M-F 375 4WD  
 5008 M-F 390 2WD  
 5009 M-F 390 4WD  
 5010 M-F 398 2WD  
 5011 M-F 398 4WD  
 5266 M-F 383 2WD

5276 M-F 365 2WD  
 5271 M-F 365 4WD  
 5742 M-F 383 4WD  
 5723 M-F 390T 2WD  
 5724 M-F 390T 4WD  
 5726 M-F 382 2WD  
 5727 M-F 382 4WD

# Mf390t Repair Manual

**Lingsheng Yao**



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h 2 s o 4 n a o h n a 2 s o 4 h 2 o if it takes 27.4 ml of 0.768 M NaOH to titrate 16.7 ml of H<sub>2</sub>SO<sub>4</sub> what is the concentration of the H<sub>2</sub>SO<sub>4</sub> solution? hint: balance the equation first

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web may 28 2020 if gasoline contains 84.2% carbon by mass and has a density of 0.8205 g/ml determine the mass of carbon dioxide produced during a 500 mile trip 3.785 liters per gallon answer 1.28  $\times 10^5$  g  $\text{CO}_2$

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web gas stoichiometry worksheet please answer the following on separate paper using proper units and showing all work please note that these problems require a balanced chemical equation 1. carbon monoxide reacts with oxygen to produce carbon dioxide if 1.0 l of carbon monoxide reacts with oxygen at STP

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web solutions to gas stoichiometry extra exercises complete the following stoichiometric problems communicate your problem solving approach using internationally accepted symbols for elements quantities numbers and units

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web balancing equations and simple stoichiometry key balance the following equations  $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$   $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$  key solutions for the stoichiometry practice worksheet