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 b $\text{mg}_2 \text{aq} + \text{co}_3^{2-} \text{aq} + 2\text{h}^+ \text{aq} \rightarrow 2\text{no}_3^- \text{aq} + \text{mg}$ write the balanced equation when nitric acid reacts with calcium hydroxide to produce water and calcium nitrate

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