



Overview Chemical Bonds

Directions: All of the statements below are false as written. In the space provided, write a term or phrase that makes the statement true when it is substituted for the underlined words.

1. The properties of a compound are the same as the properties of the elements that it contains.

2. Superscript numbers in chemical formulas tell how many atoms of each element are found in a unit of compound.

3. All the noble gases except helium have 18 electrons in their outer energy level.

4. A(n) chemical formula is the force that holds atoms together in a compound.

5. An ion is a(n) neutral particle that has either more or fewer electrons than protons.

6. Oxidation numbers are written as subscripts.

7. A(n) covalent bond is the force of attraction between the opposite charges of the ions in an ionic compound.

8. The charge on a compound is always positive.

9. Equal sharing of electrons in covalent bonds results in polar molecules.

10. Only two identical atoms can share electrons unequally.

11. A binary compound contains five different elements.

12. An oxidation number tells how many protons an atom must gain, lose, or share to become stable.

13. The oxidation number of the copper(II) ion is 3+.

14. When writing chemical formulas, add superscripts so that the sum of the oxidation numbers equals ten.

15. A polyatomic ion never has a positive or negative charge.

16. The polyatomic ion SO_4^{2-} is called the sulfide ion.

Overview Chemical Bonds Answer Key

Craig H. Heller

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Glencoe Science McGraw-Hill Staff,2001-06 *Human Biology: Breathing* Craig H. Heller,1999 *Inorganic Chemistry for JEE Advanced: Part 2, 3E (Free Sample)* K. S. Verma,2022-05-19 Inorganic Chemistry for JEE Advanced Part 2 a Cengage Exam Crack Series product is designed to help aspiring engineers focus on the subject of inorganic chemistry from two standpoints To develop their caliber aptitude and attitude for the engineering field and profession To strengthen their grasp and understanding of the concepts of the subjects of study and their applicability at the grassroots level Each book in this series approaches the subject in a very conceptual and coherent manner While its illustrative solved examples facilitate easy mastering of the concepts and their applications an array of solved problems exposes the students to a variety of questions that they can expect in the examination The coverage and features of this series of books make it highly useful for all those preparing for JEE Main and Advanced and aspiring to become engineers *Princeton Review AP Chemistry Premium Prep, 26th Edition* The Princeton Review,2024-08-06 Make sure you're studying with the most up to date prep materials Look for the newest edition of this title The Princeton Review AP Chemistry Premium Prep 27th Edition ISBN 9780593518236 on sale August 2025 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original product **Princeton Review AP Chemistry Prep, 25th Edition** The Princeton Review,2023-08-01 Make sure you're studying with the most up to date prep materials Look for the newest edition of this title The Princeton Review AP Chemistry Premium Prep 26th Edition ISBN 9780593517604 on sale August 2024 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original product

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