

Periodic Table of the Elements

This table gives information about the chemical elements. Elements are grouped into eight classes according to their properties. Each class is shown in a different color. Hydrogen does not belong to any one class.

Atomic number — 6

Chemical symbol — C

Element name — Carbon

12.011

Number of electrons in each shell — 2, 4

Atomic weight — 12.011 (1 indicates atomic weight of the most stable isotope)

<div><div><div>Alkali metals</div><div>Alkaline earth metals</div><div>Transition metals</div><div>Lanthanide series</div></div><div><div>Actinide series</div><div>Other metals</div><div>Nonmetals</div><div>Noble gases</div></div></div>																		<div>12.011</div> <div>1 indicates atomic weight of the most stable isotope</div>																	
<div>1 H Hydrogen 1.00794</div>																		<div>2 He Helium 4.002602</div>																	
<div>3 Li Lithium 6.941</div> <div>4 Be Beryllium 9.012182</div>																		<div>5 B Boron 10.811</div> <div>6 C Carbon 12.011</div> <div>7 N Nitrogen 14.00643</div> <div>8 O Oxygen 15.9994</div> <div>9 F Fluorine 18.9984032</div> <div>10 Ne Neon 20.1797</div>																	
<div>11 Na Sodium 22.989768</div> <div>12 Mg Magnesium 24.305</div>																		<div>13 Al Aluminum 26.981538</div> <div>14 Si Silicon 28.0855</div> <div>15 P Phosphorus 30.973762</div> <div>16 S Sulfur 32.06</div> <div>17 Cl Chlorine 35.453</div> <div>18 Ar Argon 39.948</div>																	
<div>19 K Potassium 39.0983</div> <div>20 Ca Calcium 40.078</div>																		<div>21 Sc Scandium 44.95591</div> <div>22 Ti Titanium 47.88</div> <div>23 V Vanadium 50.9415</div> <div>24 Cr Chromium 51.9961</div> <div>25 Mn Manganese 54.938045</div> <div>26 Fe Iron 55.845</div> <div>27 Co Cobalt 58.933195</div> <div>28 Ni Nickel 58.6934</div> <div>29 Cu Copper 63.546</div> <div>30 Zn Zinc 65.38</div> <div>31 Ga Gallium 69.723</div> <div>32 Ge Germanium 72.63</div> <div>33 As Arsenic 74.9216</div> <div>34 Se Selenium 78.96</div> <div>35 Br Bromine 79.904</div> <div>36 Kr Krypton 83.80</div>																	
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<div>87 Fr Francium 223</div> <div>88 Ra Radium 226</div>																		<div>89 Ac Actinium 227</div> <div>90 Th Thorium 232.0377</div> <div>91 Pa Protactinium 231.03688</div> <div>92 U Uranium 238.02891</div> <div>93 Np Neptunium 237</div> <div>94 Pu Plutonium 244</div> <div>95 Am Americium 243</div> <div>96 Cm Curium 247</div> <div>97 Bk Berkelium 247</div> <div>98 Cf Californium 251</div> <div>99 Es Einsteinium 252</div> <div>100 Fm Fermium 257</div> <div>101 Md Mendelevium 258</div> <div>102 No Nobelium 259</div> <div>103 Lr Lawrencium 262</div> <div>104 Rf Rutherfordium 261</div> <div>105 Db Dubnium 262</div> <div>106 Sg Seaborgium 266</div> <div>107 Bh Bohrium 264</div> <div>108 Hs Hassium 277</div> <div>109 Mt Meitnerium 268</div> <div>110 Ds Darmstadtium 271</div> <div>111 Rg Roentgenium 272</div> <div>112 Uub Ununbium 285</div> <div>113 Uut Ununtrium 284</div> <div>114 Uuq Ununquadium 289</div> <div>115 Uup Ununpentium 288</div> <div>116 Uuh Ununhexium 292</div> <div>117 Uus Ununseptium 294</div> <div>118 Uuo Ununoctium 294</div> <div>119 Uuh Ununseptium 294</div> <div>120 Uuo Ununoctium 294</div>																	

Note: Elements 112, 113, and 114 are not currently known.

Note: The adjacent numbers to those reported in tables by International Union of Pure and Applied Chemistry (IUPAC) are the accepted values in the field of chemistry. It is the business of scientists and states to make.

57 La Lanthanum 138.90547	58 Ce Cerium 140.12	59 Pr Praseodymium 140.90768	60 Nd Neodymium 144.24	61 Pm Promethium 145	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92534	66 Dy Dysprosium 162.5	67 Ho Holmium 164.93033	68 Er Erbium 167.26	69 Tm Thulium 168.93421	70 Yb Ytterbium 173.054	71 Lu Lutetium 174.967
89 Ac Actinium 227	90 Th Thorium 232.0377	91 Pa Protactinium 231.03688	92 U Uranium 238.02891	93 Np Neptunium 237	94 Pu Plutonium 244	95 Am Americium 243	96 Cm Curium 247	97 Bk Berkelium 247	98 Cf Californium 251	99 Es Einsteinium 252	100 Fm Fermium 257	101 Md Mendelevium 258	102 No Nobelium 259	103 Lr Lawrencium 262

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Didier Musso



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