

Name: _____

Atomic Structure & the Periodic Table Study Guide

Atomic Structure

Particle	Charge	Where are they found?
Proton	Positive	Nucleus
Neutron	Neutral/ no charge	Nucleus
Electron	negative	Electron cloud/energy levels outside nucleus

1. What is the rule for writing chemical symbols like Na or Ca? **Capitalize the first letter; lower case second letter**
2. What are the vertical columns on the periodic table called? Groups or families
 - a. How many of these are on the periodic table? 18
 - b. Why are the elements in these columns? They have the same chemical properties (& same # of valence electrons)
 - c. Elements in the same group have the same number of valence electrons.
3. What are the horizontal rows on the periodic table called? periods
 - a. How many of these are on the periodic table? 7
 - b. Elements in these rows have the same number of energy (*electron shells*) levels.
4. What are valence electrons? **Electrons in the outer energy level**
5. How many electrons are in each energy level?
1st energy level 2 2nd energy level 8 3rd energy level 8
6. The atomic number is equal to the # of protons and also # of electrons
7. The atomic mass is equal to the # protons + # of neutrons
8. How do you find the number of neutrons? Atomic mass – atomic number
9. We view the order of the periodic table today according to the atomic number

Periodic Table Study Guide Answers Key

**Barron's Educational Series, Brian
Stewart**



Periodic Table Study Guide Answers Key:

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Table of Contents Periodic Table Study Guide Answers Key

1. Understanding the eBook Periodic Table Study Guide Answers Key
 - The Rise of Digital Reading Periodic Table Study Guide Answers Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Study Guide Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Study Guide Answers Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Study Guide Answers Key
 - Personalized Recommendations
 - Periodic Table Study Guide Answers Key User Reviews and Ratings
 - Periodic Table Study Guide Answers Key and Bestseller Lists
5. Accessing Periodic Table Study Guide Answers Key Free and Paid eBooks

- Periodic Table Study Guide Answers Key Public Domain eBooks
 - Periodic Table Study Guide Answers Key eBook Subscription Services
 - Periodic Table Study Guide Answers Key Budget-Friendly Options
6. Navigating Periodic Table Study Guide Answers Key eBook Formats
- ePub, PDF, MOBI, and More
 - Periodic Table Study Guide Answers Key Compatibility with Devices
 - Periodic Table Study Guide Answers Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Periodic Table Study Guide Answers Key
 - Highlighting and Note-Taking Periodic Table Study Guide Answers Key
 - Interactive Elements Periodic Table Study Guide Answers Key
8. Staying Engaged with Periodic Table Study Guide Answers Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Study Guide Answers Key
9. Balancing eBooks and Physical Books Periodic Table Study Guide Answers Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Study Guide Answers Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Study Guide Answers Key
- Setting Reading Goals Periodic Table Study Guide Answers Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Study Guide Answers Key
- Fact-Checking eBook Content of Periodic Table Study Guide Answers Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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