

PERIODIC TABLE

(BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1 H Hydrogen 1.008	2 He Helium 4.0026																	
3 Li Lithium 6.941	4 Be Beryllium 9.0122																	
5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180													
11 Na Sodium 22.990	12 Mg Magnesium 24.305	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948											
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 52.00	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.69	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80	
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.36	47 Ag Silver 107.868	48 Cd Cadmium 112.41	49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29	
55 Cs Cesium 132.91	56 Ba Barium 137.33	57-71 Lanthanides		72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.084	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium 209	85 At Astatine 210	
87 Fr Francium 223	88 Ra Radium 226	89-103 Actinides		104 Rf Rutherfordium 261	105 Db Dubnium 262	106 Sg Seaborgium 266	107 Bh Bohrium 264	108 Hs Hassium 277	109 Mt Meitnerium 268	110 Ds Darmstadtium 271	111 Rg Roentgenium 272	112 Cn Copernicium 285	113 Nh Nihonium 286	114 Fl Flerovium 289	115 Uup Ununpentium 288	116 Lv Livermorium 293	117 Uus Ununseptium 294	118 Uuo Ununoctium 294

C Solid

Hg Liquid

H Gas

Rf Unknown

Metals

Nonmetals

Alkali metals

Alkaline earth metals

Lanthanides

Actinides

Transition metals

Post-transition metals

Other metalloids

Halogens

Noble gases

Atomic #

Symbol

Element name

Weight

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

270

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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Chemistry

(Speedy Study Guides) Speedy Publishing,2014-08-04 Chemistry is briefly the study of the transformation or various changes of matter Chemistry deals primarily with the structure of the material tangible world and also with the intangible the small microcosmic world as well Chemistry is a very difficult subject for those that have never taken it or do not have a general knowledge of it due to the proliferation of arcane terminology and mathematical formulas If you are having trouble with your chemistry class then do some extra research It will be both advantageous and rewarding There are many boxed book sets on the subject but you should try to find one that both fits your level of understanding and level of reading comprehension

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concepts to math An important aspect of this is knowing all of the terms related to chemistry and a student can make the process of learning these ideas is with the use of a study guide The study guide features practice forms that help a student to learn the ideas and it helps a student to keep focused during long study sessions *The Periodic Table of Elements - Alkali Metals, Alkaline Earth Metals and Transition Metals | Children's Chemistry Book* Baby Professor,2017-02-15 In this reference book we re going to discuss the alkali metals the alkaline Earth metals and the transition metals This is a very lively discussion that s complete with carefully selected information and pictures You can read this book on your own or share it with some of your friends Happy reading **Periodic Table (Advanced) (Speedy Study Guide)** Speedy Publishing LLC,2014-11-01 If you are studying any type of science and need to know about the periodic table and all of its elements you may benefit from a periodic table advanced study guide in order to help you organize and apply the information in a matter that makes sense With so many different elements on the table to memorize you may be overwhelmed by the amount of information Having a study guide helps to break down the information for you so that it is easier to understand and remember *Chemistry Basics (Speedy Study Guide)* Speedy Publishing LLC,2014-12-19 A study guide of chemistry basics would give students bite sized chunks of information on a complex subject making the task of learning it easier and more enjoyable Study guides present detailed information on a topic in a compact and simplistic way which helps the student grasp more of what they are being taught Also the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory *Chemistry Lab Basics (Speedy Study Guides)* Speedy Publishing LLC,2015-01-23 A study guide is an excellent foundation especially when you are pursuing knowledge in science Science is all about facts and provable information In chemistry you study a lot of compounds and combinations of information and without the building blocks you ve got nothing to work with Getting help with those harder concepts and reminding yourself of the easy ones can save your life and make it easier to pass those classes or spark a passion

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language This branch of chemistry deals with the molecular components that make up life on our planet

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