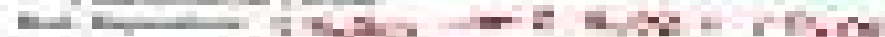


Stoichiometry Worksheet 2: Percent Yield

What aspects of a the percent yield calculation:

- Identify a balanced chemical equation of a reaction.
- Calculate the actual yield of a reaction (actual yield is what you actually get from a reaction).
- Identify the theoretical yield of a reaction (theoretical yield is what you would get if the reaction went to 100% completion).

1. Calculate the theoretical yield of a reaction (theoretical yield is what you would get if the reaction went to 100% completion).



$$\text{Theoretical yield} = \frac{\text{actual yield}}{\text{percent yield}} \times 100$$

$$\text{Theoretical yield} = \frac{\text{actual yield}}{\text{percent yield}} \times 100$$

$$\text{Theoretical yield} = \frac{\text{actual yield}}{\text{percent yield}} \times 100$$

2. Calculate the percent yield of a reaction (percent yield is what you actually get from a reaction).



$$\text{Percent yield} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$$

3. Calculate the percent yield of a reaction (percent yield is what you actually get from a reaction).



$$\text{Percent yield} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$$

4. Calculate the percent yield of a reaction (percent yield is what you actually get from a reaction).



$$\text{Percent yield} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$$

$$\text{Percent yield} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$$

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