

## Chemistry — Unit 4: Motion

### Ballon's Playground

In this case, 100% oxygen, through the use of a diver's breathing tank, allows divers to breathe underwater. However, breathing pure oxygen can be dangerous. In a simulation of this scenario:

https://www.youtube.com/watch?v=9486t8u3u00 will allow you to experience a simulation of this type of scenario. In any simulation, a variable is a quantity that can change and a constant is a quantity that does not change.

#### Part 1 – Principles

Unit	100%	200%	250%
Mass of water produced (grams)	10.00	20.00	25.00
Volume of gas (liters)	1.00	2.00	2.50

- What happened to the mass of the material in the flask as it was heated? Is this a change in mass?
- What did you notice about the volume of the gas produced and the mass of the flask after heating?
- Does the mass of the gas produced change?
- Does the relationship between the volume of gas produced and the mass of the water that was heated change?
- What are some other common reactions that result in the production of water and a gas?

#### Part 2 – Experiment

The following table shows the initial and final masses for oxygen and hydrogen in each of the trials conducted for this experiment.

	Mass 1	Mass 2	Mass 3
Mass of oxygen (g)	1.00	1.00	1.00
Mass of hydrogen (g)	0.50	0.50	0.50
Mass of product (g)	1.50	1.50	1.50
Volume of gas (liters)	1.00	1.00	1.00
Volume of hydrogen (liters)	0.50	0.50	0.50
Volume of oxygen (liters)	0.50	0.50	0.50

# Modeling Chemistry Dalton Playhouse Notes Answers

**Robert R. Redfield**



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