

## ENERGY CHANGES IN PHYSICAL AND CHEMICAL PROCESSES.

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### Objectives

By the end of this topic, the learner should be able to:

- Define endothermic and exothermic reactions using the  $\Delta H$  notation.
- Draw energy level and energy cycle diagrams.
- Explain fusion and vaporisation as evidence of inter-particle forces.
- Explain that energy changes in chemical reactions are due to bond breaking and bond formation.
- Define and explain various types of heat changes.
- Carry out some experiments to determine enthalpy changes for some reactions.
- Write correct simple thermo-chemical equations.
- State Hess's Law and carry out related calculations.
- State and explain the factors that influence the choice of a fuel.
- Explain the environmental effects of fuels.

## ENERGY CHANGES IN PHYSICAL AND CHEMICAL PROCESSES.

Energy is the ability to do work. There are many forms of energy such as chemical, electrical, heat, kinetic and potential energy.

# Mstahiki Meya Guide

**LP Steffe**



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Kenya National Bibliography Kenya National Library Service. National Reference & Bibliographic Department, 2012

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