



Test 1

Using Numbers

Q1

Order the following numbers from smallest to largest:

-2.071 , 0.2071 , 2.017 , -2.701 , -2.710

[1 mark]

Q2

Calculate the following:

$$68.52 - 11.673$$

[1 mark]

Q3

Calculate the following:

$$\frac{13}{6} - \frac{7}{4}$$

Give your answer in its simplest form.

[2 marks]

Q4

Estimate

$$\frac{12.31 \times 1.92}{4.86 + 3.07}$$

[1 mark]

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Madhubun



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assessing children's work and progress The inclusion of practical activities discussion topics photographs cartoons and case examples makes this a very user friendly book for both trainee and experienced teachers in primary and secondary schools

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Table of Contents Move On Maths Level 2 Practice Tests

1. Understanding the eBook Move On Maths Level 2 Practice Tests
 - The Rise of Digital Reading Move On Maths Level 2 Practice Tests
 - Advantages of eBooks Over Traditional Books
2. Identifying Move On Maths Level 2 Practice Tests
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Move On Maths Level 2 Practice Tests
 - User-Friendly Interface
4. Exploring eBook Recommendations from Move On Maths Level 2 Practice Tests
 - Personalized Recommendations
 - Move On Maths Level 2 Practice Tests User Reviews and Ratings
 - Move On Maths Level 2 Practice Tests and Bestseller Lists
5. Accessing Move On Maths Level 2 Practice Tests Free and Paid eBooks
 - Move On Maths Level 2 Practice Tests Public Domain eBooks
 - Move On Maths Level 2 Practice Tests eBook Subscription Services
 - Move On Maths Level 2 Practice Tests Budget-Friendly Options
6. Navigating Move On Maths Level 2 Practice Tests eBook Formats

- ePub, PDF, MOBI, and More
 - Move On Maths Level 2 Practice Tests Compatibility with Devices
 - Move On Maths Level 2 Practice Tests Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Move On Maths Level 2 Practice Tests
 - Highlighting and Note-Taking Move On Maths Level 2 Practice Tests
 - Interactive Elements Move On Maths Level 2 Practice Tests
 8. Staying Engaged with Move On Maths Level 2 Practice Tests
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Move On Maths Level 2 Practice Tests
 9. Balancing eBooks and Physical Books Move On Maths Level 2 Practice Tests
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Move On Maths Level 2 Practice Tests
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Move On Maths Level 2 Practice Tests
 - Setting Reading Goals Move On Maths Level 2 Practice Tests
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Move On Maths Level 2 Practice Tests
 - Fact-Checking eBook Content of Move On Maths Level 2 Practice Tests
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web the thermal energy is the average potential energy of the particles in a system the thermal energy is the total sum of the potential energies of the particles in a system the thermal energy is the average kinetic energy of the particles due to the interaction among the particles in a system

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web thermal energy the molecules in the air begin to move faster hence them having a higher temperature as the molecules move faster they begin to rise note that the tape will act like part of the shape and so we recommend taping it on the back 6 put both pieces of paper out in the sun 7 after two minutes collect the solar paper

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web thermal energy from the hotter object to the cooler object the hotter object will cool down and the cooler one will warm up until they are at the same temperature where there will be no net energy transfer two objects are in thermal equilibrium when there is no net transfer of thermal energy between them solids liquids and gases

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web thermal energy is the thing that object possess but heat is a process that occurs between objects if we compare an iceberg to a pot of boiling water each of the molecules of both objects have their own amount of thermal energy all pieces of matter even very cold ones have at least some thermal energy because they are in motion at the

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web the thermal energy of an object changes when heat flows into or out of the object you can use the following equation to calculate the change in thermal energy

$$\Delta Q = mc\Delta T$$
 where ΔQ is the change in thermal energy in joules (J), m is the mass in kilograms (kg), c is the specific heat capacity in J kg⁻¹ °C⁻¹, and ΔT is the change in temperature in °C.

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web thermal energy is defined as when any atoms or molecules of an element start vibrating due to the rise in temperature it produces energy which is called thermal energy in other words thermal energy definition can be the movement of atoms and molecules these atoms and molecules will keep moving faster as the temperature rises

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web the term thermal energy is used loosely in various contexts in physics and engineering generally related to the kinetic energy of vibrating and colliding atoms in a substance it can refer to several different well defined physical concepts

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