

## Chapter 19

1. a) Contains as many hydrogen atoms as possible for the given number of carbons. Only carbon-carbon single bonds.



(ii) liquid



2. a) a carbon-carbon double bond



(Home schooling parents: It doesn't matter what position you draw the bromine in on the right-hand carbon - N, S or E (with hydrogens in the other two positions). All that matters is that one bromine becomes attached to each of the carbons either side of the original double bond.)



(ii) In this (substitution) reaction, one of the hydrogens has been replaced by a bromine during the reaction. In the previous (addition) reaction, nothing was lost when the two molecules combined together.

3. a) Ethene is passed with steam at  $300^\circ\text{C}$  and 60 - 70 atmospheres over a phosphoric acid catalyst.

b) e.g. two of: continuous flow process, therefore faster; rapid reaction; purer product.

c) uses renewal resources; gentle conditions.

4. Dissolve the sugar in water in a conical flask (or similar) and add yeast. You now need some means of keeping the air out while allowing carbon dioxide to escape. You could use a commercial home-winemaking airlock of some sort, but a plug of cotton wool works just as well. Leave the mixture in the warm for a few days until bubbling stops.



Allow the yeast to settle, and pour off the solution so that it can be fractionally distilled, collecting the fraction which boils at  $78^\circ\text{C}$ . (See the diagram for the fractional distillation of an alcohol/water mixture in Chapter 11 of the book.)

continued . . .

# Pearson Answer Manual

**Nivaldo Tro, Mary Beth  
Kramer, Kathleen Shaginaw**



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1. Understanding the eBook Pearson Answer Manual
  - The Rise of Digital Reading Pearson Answer Manual
  - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Answer Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pearson Answer Manual
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Answer Manual
  - Personalized Recommendations
  - Pearson Answer Manual User Reviews and Ratings
  - Pearson Answer Manual and Bestseller Lists
5. Accessing Pearson Answer Manual Free and Paid eBooks

- Pearson Answer Manual Public Domain eBooks
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  - Pearson Answer Manual Budget-Friendly Options
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  - Pearson Answer Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Pearson Answer Manual
  - Highlighting and Note-Taking Pearson Answer Manual
  - Interactive Elements Pearson Answer Manual
8. Staying Engaged with Pearson Answer Manual
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pearson Answer Manual
9. Balancing eBooks and Physical Books Pearson Answer Manual
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pearson Answer Manual
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Pearson Answer Manual
- Setting Reading Goals Pearson Answer Manual
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Answer Manual
- Fact-Checking eBook Content of Pearson Answer Manual
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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