



# Minn Kota Terrova User Manual

**Roman Wölfel**



## **Minn Kota Terrova User Manual:**

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$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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web  $\text{cu clo}_4$  2 is soluble all perchlorates are soluble while  $\text{pbso}_4$  is insoluble sulfates are generally soluble except for  $\text{pb}^{2+}$   $\text{ba}^{2+}$   $\text{hg}_2^{2+}$  and  $\text{sr}^{2+}$  the net ionic equation is

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