

Name _____ Date _____ Class _____

CHAPTER

3

STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
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Matter is anything with (1) mass and volume. A(2) substance is a form of matter with a uniform and unchanging composition.Substances have specific, unchanging (3) properties that can be observed.Substances have both physical and chemical properties. (4) Physical

properties can be observed without changing a substance's chemical composition. Color,

hardness, and (5) density are examples. Other properties cannot be

observed without changing the composition of a substance. These are called

(6) chemical properties. An example is the tendency of iron to form

rust when exposed to air.

Label each property as either physical or chemical.

- physical 7. Chemical formula H_2O
- chemical 8. Irons react vigorously when exposed to moist air
- chemical 9. Remains unchanged when in the presence of nitrogen
- physical 10. Colorless
- physical 11. Solid at normal temperatures and pressures
- chemical 12. Ability to combine with another substance
- physical 13. Melting point
- physical 14. Liquid at normal temperatures and pressures
- physical 15. Boiling point is 100°C
- physical 16. Conducts electricity
- physical 17. Density is $\frac{1}{2} \frac{\text{g}}{\text{cm}^3}$

Name _____ Date _____ Class _____

CHAPTER

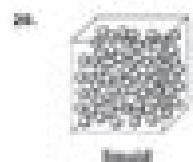
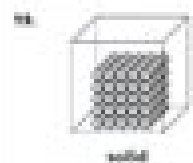
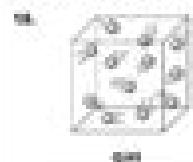
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STUDY GUIDE FOR CONTENT MASTERY

Section 3.1 continued

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- true 21. All matter that we encounter in everyday life exists in one of these physical forms.
- true 22. A solid has definite shape and volume.
- false 23. A liquid has a definite shape and takes on the volume of its container.
- true 24. A gas has both the shape and the volume of its container.
- false 25. The particles in a gas cannot be compressed into a smaller volume.
- false 26. Liquids tend to contract when heated.
- false 27. The particles in a solid are spaced far apart.
- false 28. The words gas and vapor can be used interchangeably.

Mixtures And Solution Study Guide Answers

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Mixtures And Solution Study Guide Answers:

Excel Science Study Guide, Years 7-8 Nicholas Pefani, 2005 **Study Material Based On NCERT Science Class - IX** Dr. Sunita Bhagiya, ,Er. Meera Goyal, 2022-02-16 1 Matter In Our Surrounding 2 Is Matter Around us Pure 3 Atoms And Molecules 4 Structure of the atoms 5 The Fundamental Unit of life 6 Tissues 7 Diversity in Living Organisms 8 Motion 9 Force and Laws of Motion 10 Gravitation 11 Work And Energy 12 Sound 13 Why Do we Fall Ill 14 Natural Resources 15 Improvement in Food resources Practical Work Project Work **Study Guide for The Anatomy and Physiology Learning System** Edith Applegate, 2010-03-30 Designed to accompany The Anatomy and Physiology Learning System 4th Edition by Edith Applegate this study guide helps you learn and review basic A P concepts Each chapter emphasizes medical terminology with a set of key terms word parts clinical terms and abbreviations and then adds a variety of fun filled learning exercises review questions a quiz and a word puzzle The study guide corresponds to the textbook chapter for chapter Chapter learning objectives help you focus on the most important material Key concepts are defined on the first page of each chapter in the workbook Learning exercises for each chapter include short answer matching and diagrams to label and color Self quizzes allow you to measure your progress and understanding Fun and Games features end each chapter with a variety of engaging puzzles covering words and concepts A chapter summary provides a brief review of each chapter A chapter review provides questions for reinforcement and review of the concepts in each chapter **Study Guide/Selected Solutions Manual** Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **Review Guide for RN Pre-Entrance Exam** National League for Nursing, National League for Nursing. Testing Division, 2009-09-29 One CD ROM disc in pocket **Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today** Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Denniston Katherine, 2006

The Biology Teacher's Survival Guide Michael F. Fleming, 2015-04-01 This unique resource is packed with novel and innovative ideas and activities you can put to use immediately to enliven and enrich your teaching of biology streamline your classroom management and free up your time to accomplish the many other tasks teachers constantly face For easy use materials are printed in a big 8 x 11 lay flat binding that opens flat for photo copying of evaluation forms and student activity sheets and are organized into five distinct sections 1 Innovative Classroom Techniques for the Teacher presents technique to help you stimulate active students participation in the learning process including an alternative to written exams ways to

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