Chemistry 530 Midterm Review Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- · Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section III - 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - o Percent Composition
 - Atomic Mass.
 - Electron configuration
 - Nomenclature
 - o Balancing Equations
 - Predicting the products of a reactions.
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter I – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - All non-zero digits are significant.
 - 2) Zeros are significant only when they are between two non-zero digits
 - or if they occur after a decimal point and after a non-zero digit.
 - 3) In multiplication or division, you cannot end up with more significant figures than you started with.
 - 4) In addition or subtraction, you cannot have more digits to the right of the decimal than you started with.
- Exponential numbers
- Scientific Notation = D.DD x 10° where 1.00 ≤ D.DD ≤ 9.99

Modern Chemistry Midterm Study Guide 100 Questions

Lei Shi

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