



INSTRUCTION MANUAL

FOR
BETTER OPERATION AND MAINTENANCE
OF
BED TYPE MILLING MACHINE
Model: MH-2V & MH-2P

Set. # 308

OSAKA KIKO CO., LTD.

MACHINE TOOL DIVISION

Osaka, Japan

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 eg 1 hcl aq + naoh aq → nacl aq + h₂o l for which Δh = 57 kJ mol⁻¹ eg 2 hno₃ aq + koh aq → kno₃ aq + h₂o l for which Δh = 57 kJ mol⁻¹

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o aq}$ you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts