

# ONKYO® SERVICE MANUAL

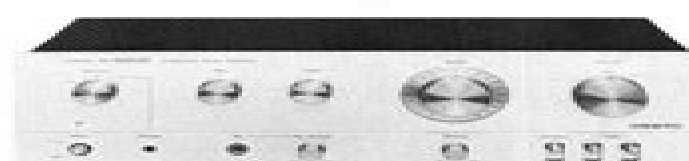
## TURNTABLE SYSTEM CP-5000A



## AM/FM STEREO TUNER T-5000



## INTEGRATED STEREO AMPLIFIER A-3000



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**R Pring**



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web the hardness of water hardness count is usually expressed in parts per million by mass of  $\text{CaCO}_3$  which is equivalent to milligrams of  $\text{CaCO}_3$  per liter of water what is the molar concentration of  $\text{Ca}^{2+}$  ions in a water sample with a hardness count of 175  $\text{mg CaCO}_3 \text{ l}^{-1}$

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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it is written like this  $\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$

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web aug 13 2017 molarity practice problems this chemistry video tutorial explains how to solve common molarity problems it discusses how to calculate the concentration of a solution given the mass in grams