



# Mercury 60 Hp Bigfoot 2015 Manual

**Mike Jess**



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web the molar volume will be the volume occupied by one molar mass  $32 \text{ g}$  of the liquid expressing the density in liters instead of ml we have  $v = \frac{m}{d}$   $\frac{32 \text{ g}}{1.790 \text{ g/l}} = 17.88 \text{ l}$   $\frac{1.0405 \text{ l}}{1 \text{ mol}} = 1.0405 \text{ l/mol}$  nonumber

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web zero degrees celsius or 273 k is defined as standard temperature and one atmosphere or 760 mm hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure stp is  $22.4 \text{ l}$

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web at standard temperature and pressure stp the molar volume  $V_m$  is the volume occupied by one mole of a chemical element or a chemical compound it can be calculated by dividing the molar mass  $M$  by mass density  $\rho$  molar gas volume is one mole of any gas at a specific temperature and pressure has a fixed volume molar volume formula

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