

Methyl Bromide: The Green's Enemy

Greenhouse gases (GHGs) are pollutants resulting from human activities that trap heat in the atmosphere and thus contribute to global warming. Among the most significant greenhouse gases is carbon dioxide (CO₂), which is released through the burning of fossil fuels. Another major greenhouse gas is methane (CH₄), which is produced by the decomposition of organic matter in landfills and the digestive processes of ruminant animals. A third significant greenhouse gas is hydrofluorocarbons (HFCs), which are synthetic compounds used in a variety of industrial and commercial applications. HFCs are highly effective at trapping heat in the atmosphere and are also long-lived, meaning they can remain in the atmosphere for decades or even centuries. The production and use of HFCs is a major source of greenhouse gas emissions, and it is one of the fastest-growing sources of GHGs. The Intergovernmental Panel on Climate Change (IPCC) has estimated that HFC emissions will increase by 50% or more by 2050. The production and use of HFCs is a major source of greenhouse gas emissions, and it is one of the fastest-growing sources of GHGs. The Intergovernmental Panel on Climate Change (IPCC) has estimated that HFC emissions will increase by 50% or more by 2050. The production and use of HFCs is a major source of greenhouse gas emissions, and it is one of the fastest-growing sources of GHGs. The Intergovernmental Panel on Climate Change (IPCC) has estimated that HFC emissions will increase by 50% or more by 2050.

In this lab, you will study the chemical structure and physical properties of methyl bromide, a potent greenhouse gas. You will also learn about the environmental impacts of methyl bromide and the methods used to control its emissions.

Objectives: Upon completion of this lab, you should be able to:

Describe the chemical structure and physical properties of methyl bromide.

Materials: 1. Methyl bromide (CH₃Br) 2. Hydrogen bromide (HBr) 3. Sodium hydroxide (NaOH) 4. Ethanol (C₂H₅OH) 5. Water (H₂O) 6. Ice (H₂O) 7. Thermometer 8. Beaker 9. Test tube 10. Dropper 11. Safety goggles 12. Lab coat 13. Gloves 14. Mask 15. First aid kit

Procedure:

1. Weigh out 10.0 g of methyl bromide (CH₃Br) and place it in a beaker. Add 10.0 g of water (H₂O) to the beaker. Stir the mixture with a glass rod. Observe the color change. Record your observations.
2. Weigh out 10.0 g of hydrogen bromide (HBr) and place it in a beaker. Add 10.0 g of water (H₂O) to the beaker. Stir the mixture with a glass rod. Observe the color change. Record your observations.
3. Weigh out 10.0 g of sodium hydroxide (NaOH) and place it in a beaker. Add 10.0 g of water (H₂O) to the beaker. Stir the mixture with a glass rod. Observe the color change. Record your observations.
4. Weigh out 10.0 g of ethanol (C₂H₅OH) and place it in a beaker. Add 10.0 g of water (H₂O) to the beaker. Stir the mixture with a glass rod. Observe the color change. Record your observations.
5. Weigh out 10.0 g of ice (H₂O) and place it in a beaker. Add 10.0 g of water (H₂O) to the beaker. Stir the mixture with a glass rod. Observe the color change. Record your observations.

Methyl Bromide The Ozone's Enemy Lab Answers

JE Gale



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Agriculture, Rural Development, and Related Agencies Appropriations for Fiscal Year 1999 United States. Congress. Senate. Committee on Appropriations. Subcommittee on Agriculture, Rural Development, and Related Agencies, 1999

Stored Product Protection E. Highley, 1994 The International Working Conference on Stored Product Protection IWCSPP held every four years is the premier world forum for presentation of research results bearing on the safe storage of durable foodstuffs particularly cereal grains pulses and oilseeds The 6th IWCSPP was held in Canberra Australia from 17-23 April 1994 under the auspices of CSIRO the Australian Centre for International Agricultural Research ACIAR and the Australian grain handling and storage industry These two volumes present the proceedings of the conference There were over 400 participants Apart from Australia there were substantial contingents from Brazil Canada China France Germany India Indonesia Israel Pakistan the Philippines the UK and the USA A further 20 countries were represented The overall objective was to present advances in research and development on the preservation of stored durable commodities since the 5th IWCSPP held in Bordeaux France 1990 Much had happened in the intervening four years the role of methyl bromide had come under scrutiny pressure from consumers and environmentalists to reduce chemical additives to foodstuffs had become more intense that mycotoxin contamination of food and feedstuffs was a serious threat to human and animal health had gained general acceptance Recognizing these and related issues the theme Stored product protection a Time of Challenge was adopted for the conference Some 250 papers were presented These proceedings volumes contain in addition to most of those papers reports of a series of workshops on specialist topics trapping application of inert dusts extension and small scale and farm storage standards expert systems and appropriate storage that were also part of the scientific program Overall the volumes provide an essential update for entomologists plant pathologists engineers and others concerned with stored product protection Agriculture, Rural Development, Food and Drug Administration, and Related Agencies

Appropriations for 2015 United States. Congress. House. Committee on Appropriations. Subcommittee on Agriculture, Rural Development, Food and Drug Administration, and Related Agencies, 2014

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Agencies, 1993 **California Farmer**, 1994 **JIRCAS Working Report**, 2006 **Pesticides Documentation Bulletin**, 1968 **F&S Index United States Annual**, 1996 Annual Report Kokusai Nōrin Suisangyō Kenkyū Sentā, 1995 **ESA**

Newsletter, 1999 Environment Reporter, 2000 **Development of Low-input Technology for Reducing**

Postharvest Losses of Staples in Southeast Asia Yutaka Mori, 2006 Survey of postharvest losses of rice and identification

of causes of quality change Analysis of annual incidence of major stored products insects and mechanisms of damage and possible prevention Development of low input drying technology and biological control of stored products insects using natural enemies and products Development of environment safe technologies for reducing postharvest losses of staples

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beneath the anus to form a small projection called the genital papilla a male pig will possess a urogenital opening for
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shaped ovaries located just posterior to the kidneys and connected to the curly oviducts these typically are quite small in the

fetal pig trace the oviducts toward the posterior to find that they merge at the uterus

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web obtain a fetal pig and identify the structures listed in figure 1 use figures 1 4 below to identify its sex use your pig and

also a pig of the opposite sex to identify the structures in the photographs below the word urogenital refers to an opening

that serves both the urinary excretory and the reproductive systems figure 1

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web 1 most of the pig's external features are familiar to you ears nose eyes etc on the belly you will see the umbilical cord

which connected the fetal pig to its mother's placenta on either side of the umbilical cord you may see mammary papillae

little nipples that will turn into teats in female pigs 2

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